NOTE

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New lignan, isoactifolin, from Chamaecyparis obtusa cv. Breviramea

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Abstract A new lignan isoactifolin was isolated from young shoots (with leaves) of Chamaecyparis obtusa cv. Breviramea. The structure of the compound was determined based on spectroscopic evidence.

Key words Lignan · Isoactifolin · Chamaecyparis obtusa cv. Breviramea · Furan

young leaves, a detailed survey of lignans, especially possible biosynthetic precursor lignans of 2 in this species, has not yet been reported. Therefore, we surveyed lignans in C. obtusa and C. obtusa cv. Breviramea and isolated 11 lignans. One was found to be a new furan lignan, isoactifolin (1), and here we report its isolation. Characterization of the other 10 known lignans will be reported elsewhere in relation to stereochemistry.

Introduction

Hinoki cypress (Chamaecyparis obtusa) has long been utilized as one of the most important building woods in Japan. Heartwood of this species contains significant amounts of a dibenzylbutyrolactone lignan, hinokinin (2).¹² A few other lignans were isolated from this and related species. Thus, 2 accounted for about 30% of the resins extracted from C. obtusa heartwood.¹ Savinin (3) (= hibalactone), a dehydro derivative of 2, was isolated from heartwood of Chamaecyparis pisifera³ and from young leaves of C. obtusa,⁴ C. obtusa cv. Breviramea,^{5,6} and C. pisifera cv. Plumosa-aurea.^{5.6} (+)-Sesamin (4) and deoxypodophyllotoxin (5) were obtained from C. obtusa cv. Breviramea young leaves⁷ and *Chamaecyparis lawsoniana* leaves,⁸ respectively. Despite the high content of 2 in C. obtusa heartwood and of 3 in C. obtusa cv. Breviramea

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Experimental

Instruments and chromatography

One- and two-dimensional nuclear magnetic resonance (NMR) spectra were obtained with a JNM-LA400MK FT-NMR system (JEOL). Chemical shifts and coupling constants (J) were expressed in δ values and Hz, respectively. Low- and high-resolution electron impact mass spectrometry (EIMS) was performed on a JMS-DX303HF mass spectrometer (JEOL) equipped with a JMA-DA5000 mass data system. High-performance liquid chromatography (HPLC) was conducted with a Shimadzu LC-6A liquid chromatograph, detection being at $\lambda = 280$ nm. The reversed-phase column used was a Waters Novapak C₁₈ $(150 \times 3.9 \text{ mm})$, and it was a eluted with CH₃CN-H₂O (37:63) at 1 ml/min. Silica gel column chromatography employed Kieselgel 60 (Merck, 70-230 mesh). Silica gel thin-layer chromatography (TLC) employed Kieselgel 60 F_{254} (Merck, 20 \times 20 cm, 0.5 or 0.25 mm). All chemicals used were of reagent grade.

Plant material

Chamaecyparis obtusa cv. Breviramea plants were obtained from a local nursery and were maintained in the experimental forest of Wood Research Institute, Kyoto University, Japan. Young shoots with leaves of the plant were collected in February 1994 and April 1999 and were used for lignan extraction.

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Isolation and structural determination of isoactifolin (1)

Freeze-dried *C. obtusa* cv. Breviramea young shoots with leaves (163.43g) were pulverized using a Waring blender and then extracted with hot methanol (400, 150, 150, 150, 150, 150, and 150 ml; total 1300 ml). The combined methanol extracts (43.76g) were suspended in distilled water (300 ml), which was then extracted with diethyl ether (200 ml \times 3). The combined diethyl ether extracts (13.26g) were submitted to successive purification by repeated column chromatography (solvents: first, EtOAc; second, methanol/dichloromethane = 3:97; third, ethyl acetate/*n*hexane = 1:1), repeated TLC (solvents: first, acetone/ dichloromethane = 3:97; second, ethyl acetate/*n*-hexane = 1:3; third, acetone/dichloromethane = 3:97), and reversedphase HPLC to afford a new lignan (1) (3.6 mg).

Isoactifolin (1): data of ¹H- and ¹³C-NMR, ¹H-¹H correlated spectrometry (COSY), and ¹H-detected heteronuclear multiple-bond quantum correlation (HMBC) are summarized in Table 1. NOE difference ¹H-NMR spectral data are shown in Table 2. MS m/z [rel. int. (%)]: 400 (76.4, M⁺), 340 (22.1), 219 (27.6), 205 (38.7), 188 (25.1), 173 (23.2), 162 (16.5), 151 (65.1), 137 (30.9), 135 (100.0), 131 (16.9), 77 (19.5). High-resolution MS m/z (M⁺): calculated for $C_{22}H_{24}O_7$: 400.1522; found: 400.1525.

Results and discussion

Because young leaves of *C. obtusa* cv. Breviramea contain significant amounts of 3,⁴⁻⁶ the plant material was subjected to a survey for lignans, especially possible biosynthetic precursors of 2 and 3. As a result, a new lignan (1) (Fig. 1) was isolated from methanol extracts of the young shoots (with leaves) by chromatographic methods. In addition to 1, several known lignans were isolated, and their characterization including stereochemical properties will be reported elsewhere.

The structure of 1 was determined by mass spectrometry and one- and two-dimensional NMR spectrometry. Compound 1 afforded a molecular ion $[M^+]$ at m/z 400.1525 in the high-resolution EIMS, indicating the molecular formula to be $C_{22}H_{24}O_7$ (calculated for $C_{22}H_{24}O_7$: 400.1522). Table 1 shows the ¹H-NMR spectral data, which displayed the presence of an aromatic methoxyl group at δ 3.89 and an aromatic methylenedioxy group at δ 5.93. In addition, it revealed six aromatic protons at δ 6.62–6.87, which showed typical patterns of 1,3,4-tri-substituted benzenes. These results suggested that 1 had 3-methoxy-4-hydroxyphenyl (guaiacyl) and 3,4-methylenedioxyphenyl (piperonyl), which are popular aromatic moieties in lignans. It also exhibited a singlet of a phenolic hydroxyl group at δ 5.56 and a singlet of an alcoholic acetyl group at δ 2.02. Other signals in the ¹H-NMR spectrum were assigned with ¹H-¹H COSY (Table 1). It was revealed that a doublet at δ 4.74 (J = 6.6, H-7) was coupled with a multiplet at δ 2.49–2.56 (H-8), which was in turn coupled with two double doublets at δ 4.16 (J = 7.3, J = 11.2, H-9) and 4.31 (J = 7.1, J = 11.2, H-9)9). The multiplet at δ 2.49–2.56 (H-8) was coupled with a multiplet at δ 2.65–2.74 (H-8'). It was also revealed that the multiplet at δ 2.65–2.74 (H-8') was coupled with four double doublets at δ 2.51 (J = 10.7, J = 13.4, H-7'), 2.81 (J= 5.0, J = 13.5, H-7', 3.70 (J = 6.7, J = 8.7, H-9'a), and 4.05(J = 6.6, J = 8.8, H-9'b). Assignments of H-9'a and H-9'b were done by nuclear Overhauser effect (NOE) difference ¹H-NMR spectrometry (Table 2). In the ¹³C-NMR spectrum, 22 carbon signals were observed, as shown in Table 1. The assignments of the carbon atoms were achieved by a combination of ¹H-detected heteronuclear multiple quantum coherence (HMQC) spectrometry and HMBC spectrometry (Table 1). These ¹H-NMR and ¹³C-NMR spectra were similar to those of an acetylated furan lignan, actifolin (6), which was isolated from Actinodaphne longifolia.⁹ However, the HMBC correlations between benzylic protons (H-7 and H-7') and aromatic carbons (Table 1) indicated that 1 is the regioisomer of 6 in terms of the positions of the aromatic rings. The result was further confirmed by EIMS. Thus, intense fragments at m/z 135 and 151 were observed in the EIMS spectrum, which were assigned to 3,4methylenedioxybenzyl ion and methoxyhydroxybenzyloxy ion, respectively, indicating that guaiacyl substituent and piperonyl substituent attached to C-7 and C-7', respectively. As for the location of the acetyl group, H-9 (δ 4.16)





Table 1. NMR data for isoactifolin (1) in CDCl₃

Carbon no.	¹³ C ^a	¹ H ^a	¹ H- ¹ H COSY ^b	HMBC ^e
1	134.26			H-5
2	108.29 or 108.35	6.84 (d, J = 2.0)		
3	146.59			H-5, OMe, OH
4	145.11			H-2, 5, 6, OH
5	114.23	6.87 (d, J = 8.1)	H-6	OH
6	118.86	$6.79 (\mathrm{dd}, J = 2.0, 8.1)$	H-5	H-2, 7
7	83.11	4.74 (d, J = 6.6)	H-8	H-2, 6, 9, 9, 9′b
8	48.96	2.49–2.56 (m)	H-7, 8', 9, 9	H-9, 9
9	62.67	$4.16 (\mathrm{dd}, J = 7.3, 11.2)$	H-8, 9	H-7, 8
		4.31 (dd, $J = 7.1, 11.2$)	H-8, 9	
1'	133.78			H-5′, 7′, 7′
2'	108.89	6.66 (d, $J = 1.7$)		H-6', 7', 7'
3'	147.82			H-2', 5', OCH ₂ O
4'	146.04			H-2′, 6′, OCH ₂ O
5'	108.35 or 108.29	6.73 (d, J = 8.1)	H- 6′	
6'	121.48	$6.62 (\mathrm{dd}, J = 1.7, 8.1)$	H-5'	H-2', 7', 7'
7'	33.30	2.51 (dd, J = 10.7, 13.4)	H-7', 8'	H-2', 6'
		2.81 (dd, $J = 5.0, 13.5$)	H-7′, 8′	
8'	42.46	2.65-2.74 (m)	H-7′, 7′, 8, 9′a, 9′b	H-7′, 9, 9
9'	72.67	$3.70 (\mathrm{dd}, J = 6.7, 8.7, \mathrm{H}-9'\mathrm{a})$	H-8′, H-9′b	H-7′
		$4.05 (\mathrm{dd}, J = 6.6, 8.8, \mathrm{H}-9'\mathrm{b})$	H-8′, H-9′a	
OMe	55.97	3.89 (s)		
OCH_2O	100.94	5.93 (s)		
OCOMe	20.92 (Me)	2.02 (s)		
	170.97(C=O)	· ·		H-9, 9, OCOMe
OH	、 ,	5.56 (s)		

^a Chemical shifts are δ values; coupling constants (J in parentheses) are given in Hz

^bCorrelations between H-2 and H-6, and between H-2' and H-6' were not observed clearly

^cProtons correlating with carbon resonances

 Table 2. Nuclear Overhauser effect difference ¹H-NMR spectral data for isoactifolin (1)

Irradiation (δ)	Enhanced peaks (δ)
2.70 (H-8')	2.51, 2.49–2.56, 2.81, 4.05, 6.62, 6.66
3.70 (H-9'a)	2.65-2.74, 4.05, 4.74, 6.62, 6.66
4.05 (H-9'b)	2.49-2.56, 2.65-2.74, 3.70, 6.62, 6.66
4.74 (H-7)	2.49-2.56, 3.70, 4.16, 4.31, 6.62, 6.66, 6.73, 6.79, 6.84

and H-9 (δ 4.31) showed a correlation with the carbonyl carbon (δ 170.97) in the HMBC spectrum (Table 1). In addition, the chemical shift of acetyl protons (δ 2.02) indicated its alcoholic, but not phenolic, acetyl nature, which was further confirmed by the downfield shift of H-9 signals (δ 4.16 and δ 4.31) compared with those of non-acetylated lariciresinol (**7**) (H-9, δ 3.77 and δ 3.92).¹⁰ These results clearly indicated that the acetyl group was attached to the alcoholic hydroxyl group on C-9.

The NOE difference ¹H-NMR spectra of **1** (Table 2) indicated that the configurations of **1** were 8,8'-*cis* and 7,8-*trans* ($7R^*, 8S^*, 8'S^*$) as in **7**.¹¹ Taken together, the relative structure for **1** was determined as shown in Fig. 1, and we propose isoactifolin as the name of **1**.

Naturally occurring acetylated lignans, such as actifolin (6), are not popular, and the isolation of 1 from *C. obtusa* cv. Breviramea was another example. In addition, this was the first isolation of a furan lignan from *Chamaecyparis* plants. Because actifolin (6) showed cytotoxicity against a

small panel of human tumor cells,¹² it is of interest to examine whether **1** has cytotoxicity.

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- 496
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